2-[(2,3-Dihydro-1*H*-indol-1-yl)methyl]melatonin Analogues: A Novel Class of MT₂-Selective Melatonin Receptor Antagonists

Darius P. Zlotos,**[†] Mohamed I. Attia,[†] Justin Julius,[§] Shalini Sethi,[§] and Paula A. Witt-Enderby[§]

Institute of Pharmacy and Food Chemistry, Pharmaceutical Chemistry, University of Würzburg, Am Hubland, 97074 Würzburg, Germany, and Division of Pharmaceutical Sciences, School of Pharmacy, Duquesne University, 421 Mellon Hall, Pittsburgh, Pennsylvania 15282

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A novel series of 2-[(2,3-dihydro-1*H*-indol-1-yl)methyl]melatonin analogues has been prepared to probe the steric and electronic properties of the binding pocket of the MT₂ receptor accommodating the "out-ofplane" substituent of MT₂-selective antagonists. The acetamide (**6b**) bearing an usubstituted indoline moiety displayed an excellent binding affinity and selectivity toward the MT₂-subtype (MT₂, $K_i = 1$ nM; MT₁, K_i = 115 nM), behaving as a competitive antagonist. 5-Me, 5-OMe, 5-Br, 6-NH₂, and 6-NO₂ substitution of the indoline moiety reduced both MT₂ affinity and selectivity, indicating that hydrophobic interactions play a decisive role in binding the out-of-plane substituent. The cyclobutanecarboxamide (**6e**) showed a biphasic binding pattern at MT₂ receptors, indicating the presence of two MT₂ binding sites, a high affinity ($K_i = 1$ pM) and a low affinity ($K_i = 148$ nM), while MT₁ binding affinity was very low ($K_i = 1.4 \mu$ M). Functional analysis of **6e** revealed it to be an antagonist at MT₁ receptors and a partial agonist, at best, at MT₂ receptors.

Introduction

Melatonin (*N*-acetyl-5-methoxytryptamine, MLT^a, Figure 1) is a neurohormone mainly secreted by the pineal gland of mammals including humans.¹ The circadian pattern of MLT secretion is associated with the "biological clock", suggesting that MLT plays an important role in modulation of the sleep-wake cycle and of circadian rhythms in humans.² Other effects of MLT described in the literature include its antiinflammatory,³ pain modulatory,⁴ retinal,⁵ vascular,⁶ antitumor,⁷ antioxidant,⁸ stroke-protective,⁹ and neuroprotective¹⁰ properties. The physiological effects of MLT result mainly from the activation of the high affinity G-protein-coupled receptors MT₁ and MT₂. Both receptor subtypes have been found in mammals, including humans, and subsequently cloned.¹¹ While it is known that MT₁ and MT₂ receptors are expressed both centrally (suprachiasmatic nucleus, cortex, pars tuberalis, etc.) and peripherally (kidney, adipocytes, retina, blood vessels, etc.),¹² their physiological roles are not well defined. MT₁ receptors seem to be involved in the sleep promoting effects of MLT^{13,14} and in mediating vasoconstriction,¹⁵ whereas MT₂ receptors appear to play a major role in the resynchronizing activity of MLT^{13,16,17} and in mediating vasodilation.

During the past 2 decades, a great number of melatonin receptor ligands have been reported in the literature.¹⁸ The nonselective melatonin receptor agonist ramelteon has been successfully marketed in the U.S. for the treatment of insomnia.^{19,20} Other nonselective agonists, such as LY-156735²¹ or tasimelteon (VEC-162),²² are undergoing clinical trials for the treatment of sleep disorders. Agomelatine, an MT₁/MT₂ agonist and 5-HT_{2c} antagonist, is under evaluation for the treatment of major depression.²³ Melatonin receptor antagonists have been only evaluated in preclinical studies, for instance, luzindole for its antidepressant-like effects,²⁴ S22153 in circa-

dian rhythm entrainment experiments,²⁵ and ML-23 in the treatment of Parkinson's disease.²⁶ While the number of MT_1 and MT_2 selective agonists as well as MT_1 -selective antagonists is still very limited,^{18,27} several series of MT_2 -selective antagonists have been reported.^{18,28} The representative agents are displayed in Table 1.

A common structural feature in most of MT₂-selective antagonists is the presence of a lipophilic substituent located out of the plane of their core nucleus in a position corresponding to positions 1 and 2 of the indole ring in melatonin.²⁸ 3D-QSAR and docking within homology models of MT₁ and MT₂ receptors revealed that MT₂ receptors possess a hydrophobic pocket accommodating the lipophilic "out-of-plane" substituent of MT₂ antagonists.³⁷ However, there are only limited studies concerning the steric and electronic tolerance of this binding cavity.^{30,33,35,37}

In order to probe the postulated pharmacophore for MT_2 antagonists, we have recently synthesized rigid pentacyclic ligands (**1a,b**) (Figure 2) possessing an indoline moiety attached to the positions 1 and 2 of melatonin.^{38,39} The racemic compounds **1a,b** exhibited nanomolar affinity for MT_2 receptors (**1a**, $K_i = 65$ nM; **1b**, $K_i = 410$ nM) being 5-fold higher than for the MT₁ subtype (**1a**, $K_i = 320$ nM; **1b**, $K_i = 1.8 \mu$ M). The most likely explanation for the poor selectivity and moderate binding of **1a,b** is the bulkiness and/or unfavorable spatial orientation of the indoline moiety, which is not able to occupy the lipophilic binding pocket of the MT₂ receptors because of the nearly planar geometry of the dihydropyrazinodiindole ring system (Figure 3).

In this paper, we report the synthesis and pharmacological evaluation of a novel class of MT_2 -selective melatonin receptor antagonists formally obtained by opening the central sixmembered ring of **1a** (Figure 2). In contrast to the rigid ring system of **1a**, the indoline moiety of the designed compounds is conformationally flexible and thus more likely to reach the lipophilic binding pocket of the MT_2 receptors. Additionally, the indoline moiety was substituted with groups of different electronic and steric properties, such as CH_3 , OCH_3 , Br, NH_2 , and NO_2 , in order to explore the binding behavior within the lipophilic pocket accommodating the "out-of-plane" substituent.

^{*} To whom correspondence should be addressed. Phone: +49 931 888 5489. Fax: +49 931 888 5494. E-mail: zlotos@pzlc.uni-wuerzburg.de.

[†] University of Würzburg.

[§] Duquesne University.

^{*a*} Abbreviations: MLT, melatonin; MT₁, melatonin receptor subtype 1, MT₂, melatonin receptor subtype 2.



Figure 1. Melatonin and melatoninergic ligands in clinical and preclinical use.

Moreover, 2-(pyrrolidin-1-ylmethyl)melatonin, an analogue lacking a benzene ring, was prepared to complete the SARs.

Chemistry

The title compounds were prepared from the commercially available 5-methoxyindole-2-carboxylic acid according to two routes. The first synthetic sequence, displayed in Scheme 1, commenced with the condensation of 5-methoxyindole-2-carboxylic acid with 2-methylindoline and indoline using 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDCI·HCl) in dry CH₂Cl₂ to give the amides **2a,b** in moderate yields. 2-Methylindoline was obtained from 2-methylindole via NaC-NBH₃ reduction.⁴⁰ For the introduction of the ethylamine side chain we used the procedure previously applied for other ring systems.^{38,41}

Our approach involved a sequence of a Mannich reaction, quaternization of the Mannich base, substitution of the trimethylamine moiety by a cyanide, and a final reduction of the cyanomethyl group to the ethylamine moiety. Thus, aminomethylation of **2a,b** using *N,N*-dimethylmethyleneiminium iodide in chloroform afforded the Mannich bases **3a,b**. Treatment of **3a,b** with methyl iodide in dichlormethane and heating of the resulting trimethylammonium iodides with potassium cyanide and dicyclohexyl-18-crown-6 in acetonitrile provided the nitriles **4a,b**. Simultaneous nitrile and amide reduction using LiAlH₄ in diethyl ether/THF afforded the ethylamines **5a,b**, which were converted to the desired melatoninergic ligands **6a**-**e** by N-acylation using acetic anhydride (**6d**), and cyclobutanecarboxylic acid chloride (**6e**).

The key intermediate for the more convenient route toward the target compounds **6f**–**k** was 3-(cynomethyl)-5-methoxyindole-2-carboxylic acid **9**, which was prepared from methyl 5-methoxy-indole-2-carboxylate in three steps using standard procedures as outlined in Scheme 2. Briefly, the known Mannich base **7**,⁴² obtained by aminomethylation of the starting material using *N*,*N*-dimethylmethyleneiminium chloride, was N-methylated and subsequently subjected to nucleophilic substitution using potassium cyanide. The resulting cyanomethyl ester **8** was hydrolyzed to the corresponding acid **9** using LiOH in THF/H₂O.

Starting from **9**, the target compounds **6f**–**i** were prepared in a reaction sequence already applied in the first route involving condensation with the appropriate amine, simultaneous nitrile and amide reduction using LiAlH₄, followed by N-acylation of the resulting amines (Scheme 3). The differently substituted indolines were obtained from the corresponding indoles via NaCNBH₃ reduction⁴⁰ except for the commercially available 6-nitroindoline. For the synthesis of 6'-nitroindoline substituted amine **5j** we used AlH₃ (created in situ from LiAlH₄/AlCl₃) as a reducing agent in order to avoid reduction of the nitro group. The 6'-amino substituted target compound **6k** was prepared from the nitroacetamide **6j** by catalytic transfer hydrogenation using ammonium formate and Pd/C (10%) (Scheme 4).

Results and Discussion

According to the existing pharmacophore, a structural characteristic of MT_2 -selective antagonists is a lipophilic substituent located out of the plane of their core nucleus in a position corresponding to position 1 or 2 of melatonin. Docking of MT_2 selective antagonists to homology models of melatonin receptors indicated that MT_2 selectivity is due to a hydrophobic binding pocket accommodating this out-of-plane substituent that is only present in the MT_2 subtype.³⁷ Our previously reported pentacyclic ligands **1a,b** are derived from melatonin by attaching the indoline group to N1 and C2 via methylene groups. As shown in Figure 3, the indoline moiety in both enantiomers of **1b** is arranged nearly coplanar to the indole ring, being unable to reach the hydrophobic binding region of the MT_2 receptor.

Consequently, the binding affinity and MT₂ selectivity of **1b** are rather poor. In order to improve both affinity and MT₂ selectivity, we synthesized a series of compounds **6a**–**k** formally obtained from **1a**,**b** by breaking the C–N bond of the central six-membered ring as shown in Figure 2. The novel melatoninderived agents bear differently substituted *N*-methylindoline moieties in position 2, which seem to be flexible enough to reach the lipophilic pocket of the MT₂ receptors. The binding affinities of **6a**–**k** for human MT₁ and MT₂ melatonin receptors measured by competition binding analysis using the radioligand 2-[¹²⁵I]-iodomelatonin⁴³ are reported in Table 2.

The data reveal a considerable increase of binding affinity at both receptor subtypes for all new derivatives when compared to compound **1b** (MT₁, $K_i = 1800$ nM; MT₂, $K_i = 410$ nM). However, the affinity improvement was strongly dependent on the substitution pattern of the indoline group. The 5-methoxyindoline derivative **6f** exerts the highest MT_1 binding affinity $(MT_1, K_i = 5.8 \text{ nM})$ and is the only high-affinity ligand in this series showing no subtype selectivity (MT₂, $K_i = 7.1$ nM). Interestingly, this finding is not in agreement with affinity data obtained in a series of 2-benzyl-substituted benzofuran analogues with the 4-methoxybenzyl derivative being one of the most MT₂selective agents (MT₁, $K_i = 24$ nM; MT₂, $K_i = 0.50$ nM).³¹ The nonselective binding behavior of 6f is probably caused by the competition of both methoxy groups for binding at the MT₂ receptor region accommodating the methoxy group of melatonin inducing the unfavorable ligand orientation. Removal of the methoxy group from the indoline moiety of 6f generated a drastic 18-fold decrease of binding at MT₁ receptors, whereas the MT₂ affinity was increased by a factor of 6. The resulting





ligand **6b** is characterized by an outstanding pharmacological profile possessing the highest binding affinity for the MT₂ receptor ($K_i = 1.1$ nM) and an excellent affinity ratio ((K_i MT₁)/(K_i MT₂)) of 91. Introduction of a methyl group into the position adjacent to the indoline nitrogen atom of **6b**, to give **6a**, decreased MT₂ and MT₁ affinity 12.5 and 1.5 times, respectively, leading to a considerably reduced MT₂ selectivity ((K_i MT₁)/(K_i MT₂)) of approximately 11. Removal of the benzene ring from the indoline group of **6b**, to yield the *N*-methylpyrrolidine



Figure 2. Design of a novel class of melatoninergic ligands.



Figure 3. Superposition of (*R*)-**1b** (A) and (*S*)-**1b** (B) onto the selective MT₂ antagonist UCM454 (yellow structure, Table 1) in a conformation adopted in the MT₂ binding pocket.³⁷

analogue **6i**, considerably decreased MT_2 binding ($K_i = 25 \text{ nM}$) while the MT_1 affinity was unaffected ($K_i = 121 \text{ nM}$), indicating the importance of the presence of aromatic rings for ligand-receptor interactions within the hydrophobic pocket of MT_2 receptors. This is consistent with the structures of other MT_2 -selective ligands, all of them having an aromatic ring in a corresponding position (see Table 1).

In order to further probe the steric, lipophilic, and electronic requirements of the MT₂ binding pocket accommodating this aromatic substituent, we substituted the indoline benzene ring with 5-Me, 5-Br, 6-NO₂, and 6-NH₂ groups. The resulting methyl, bromo, and amino substituted agents 6g, 6h, and 6k are nonselective ligands exhibiting similar MT₁ binding constants (41, 85, and 108 nM, respectively) and drastically reduced affinity for the MT₂ receptors (15, 30, and 118 nM, respectively) when compared to the nonsubstituted parent compound 6b (MT₁ $K_i = 101 \text{ nM}, \text{MT}_2 K_i = 1.1 \text{ nM}$). The 6-nitro derivative **6j** is a moderate MT₂-selective ligand (MT₁, $K_i = 302$ nM; MT₂, K_i = 26.5 nM) having a selectivity ratio of 11.5. These findings indicate that the hydrophobic binding pocket accommodating the out-of-plane substituent is sterically restricted. Morover, the ligand receptor interactions within this binding site are likely to be of a pure hydrophobic nature, as polar groups, such as NO₂ and NH₂, reduce binding.

In order to further optimize the pharmacological profile of the most MT₂-selective agent **6b**, we modulated the acyl chain by replacing the methyl group with substituents that are often present in other MT₂-selective ligands, such as ethyl and cyclobutyl (see Table 1). Surprisingly, the propionamide **6d** displayed a 12-fold lower binding affinity for MT₂ receptors ($K_i = 14.5$ nM) and 3-fold higher affinity for MT₁ receptors ($K_i = 38$ nM) than the acetamide **6b**, resulting in decreased

Scheme 1^a



^{*a*} Reagents and conditions: (i) EDCI+HCl, CH₂Cl₂, room temp; (ii) (CH₂= NMe_2)⁺I⁻, CHCl₃, reflux; (iii) (1) MeI, CH₂Cl₂, room temp, (2) KCN, dicyclohexyl-18-crown-6, MeCN, reflux; (iv) LiAlH₄, Et₂O, THF, 0 °C to room temp; (v) respective acylation agent, Et₃N, CH₂Cl₂, room temp.





^{*a*} Reagents and conditions: (i) (CH₂= NMe_2)⁺Cl⁻, CHCl₃, reflux; (ii) (1) MeI, CH₂Cl₂, room temp, (2) KCN, dicyclohexyl-18-crown-6, MeCN, reflux; (iii) LiOH, H₂O, THF, room temp.

MT₂ selectivity. A similar reduction of MT₂ selectivity was observed in the 2-methylindoline substituted agents **6a** and **6c**. While the acetamide **6a** was 11 times more selective for MT₂ ($K_i = 14$ nM) than for MT₁ receptors ($K_i = 148$ nM), the corresponding butyramide **6c** displayed an affinity ratio of 5 (MT₁, $K_i = 41$ nM; MT₂, $K_i = 8$ nM). Interestingly, replacement of the methyl group in **6b** with a cyclobutyl, to produce **6e**, generated a biphasic pharmacological profile for MT₂ receptors ($K_{iHigh} = 1$ pM and $K_{iLow} = 148$ nM) but not for MT₁ receptors ($K_i = 1.4 \mu$ M), as shown in Figure 4.

The existence of a high-affinity and a low-affinity state has been reported for the MT_1 receptor only,⁴³ and to the best of our knowledge, this is the first time such pharmacological behavior could be detected for the MT_2 receptor.

Functional analysis of **6b** showed it to be a competitive antagonist. As shown in Figure 5, the addition of **6b** along with melatonin (0.01 pM to 10 μ M) produced a rightward shift in the melatonin curves for both MT₁ and MT₂ receptors, suggestive of competitive antagonism at both receptors. The potency of melatonin

Scheme 3^a



^{*a*} Reagents and conditions: (i) EDCI·HCl, respective indoline, CH₂Cl₂, room temp; CHCl₃, reflux; (ii) LiAlH₄, Et₂O, THF, 0 °C to room temp; (iii) respective acylation agent, Et₃N, CH₂Cl₂, room temp.

Scheme 4^a NH_2 MeO (i) (ii) NO₂ 5j 4 NHAc NHAc MeC (iii) NH NH_2 NO. 6i

 a Reagents and conditions: (i) LiAlH4, AlCl3, Et2O, THF, 0 °C to room temp; (ii) Ac2O, Et3N, CH2Cl2, room temp; (iii) HCO2NH4, 10% Pd/C, EtOH.

to inhibit forskolin-induced cAMP accumulation was 0.23 pM for the high affinity state of the MT_1 receptor and 88 pM for the low affinity state of the MT_1 receptor. The addition of 10 or 100 nM of **6b** produced a potency value of 100 and 79 pM, respectively. As for the MT_2 receptor, the potency of melatonin to inhibit forskolin-induced cAMP accumulation was 61 pM, which was lowered to 263 pM and 5.5 nM in the presence of 10 or 100 nM **6b**, respectively. No concentration-dependent inhibition of forskolin-induced cAMP accumulation occurred when **6b** (0.1 pM to 10 nM) was added alone, suggesting that it displays no intrinsic activity at either the MT_1 or MT_2 receptor (data not shown). Functional analysis of **6e** showed it to lack intrinsic activity at MT_1 receptors, as no concentration-dependent inhibition of forskolin-induced cAMP accumulation occurred at MT_1 recep-

Table 2. Binding Affinity^{*a*} of Compounds **1**–**6k** for the Human MT_1 and MT_2 Receptors Expressed in CHO Cells Obtained in Competition Radioligand Binding Assays Using 2-[¹²⁵I]Iodomelatonin



				$pK_i \pm SEM$	
compd	R	\mathbb{R}^1	Х	MT_1	MT ₂
MLT				9.34 ± 0.10	9.02 ± 0.09
1a				6.52 ± 0.02	7.22 ± 0.03
1b				5.75 ± 0.21	6.39 ± 0.27
6a	Me	Me	Н	6.83 ± 0.04	7.85 ± 0.19
6b	Н	Me	Н	6.94 ± 0.02	8.93 ± 0.05
6c	Me	<i>n</i> -Pr	Н	7.39 ± 0.02	8.09 ± 0.05
6d	Н	Et	Н	7.42 ± 0.05	7.84 ± 0.19
6e	Н	c-Bu	Н	5.85 ± 0.03	12.00 ± 0.58
					6.83 ± 0.03
6f	Н	Me	5-OMe	8.24 ± 0.16	8.15 ± 0.04
6g	Н	Me	5-Me	7.39 ± 0.02	7.83 ± 0.04
6h	Н	Me	5-Br	7.07 ± 0.04	7.52 ± 0.02
6i	Н	Me		6.92 ± 0.03	7.61 ± 0.23
6j	Н	Me	$6-NO_2$	6.51 ± 0.04	7.58 ± 0.07
6k	Н	Me	6-NH ₂	6.96 ± 0.06	6.93 ± 0.06

^{*a*} pK_i values were calculated from IC₅₀ values obtained from competitive curves according to the method of Cheng and Prusoff⁴⁴ and are the mean of three determinations.

tors. Regarding its actions at MT_2 receptors, it may display partial agonist activity, at best, at MT_2 receptors at concentrations higher than 10 nM (Figure 5).

In summary, a novel series of 2-[(2,3-dihydro-1H-indol-1vl)methyl]melatonin analogues has been prepared in order to probe the steric and electronic properties of the lipophilic binding pocket of MT₂ receptors accommodating the "out-ofplane" substituent of MT₂-selective antagonists. The best tolerated substituent attached to position 2 of melatonin is the 2-(indoline-1-yl)methyl group. Increasing the bulkiness of the methylindoline moiety by a methyl group in positions 2 and 5 as well as changing the electronic, lipophilic, and hydrogen bond donor and acceptor properties by introduction of 5-OMe, 5-Br, 6-NH₂, and 6-NO₂ substituents is detrimental to MT₂ affinity and selectivity. The results indicate a sterically restricted binding pocket with hydrophobic interactions playing a decisive role in binding the out-of-plane substituent of MT₂-selective antagonists. The acetamide 6b bearing an usubstituted indoline moiety displayed an excellent binding affinity and selectivity toward the MT₂ subtype (MT₂, $K_i = 1$ nM; MT₁, $K_i = 115$ nM), behaving as a competitive antagonist. The cyclobutane amide 6e showed the most interesting pharmacological profile. Its biphasic binding pattern at MT₂ receptors indicated the presence of two MT₂ binding sites, the high affinity one ($K_i = 1$ pM) and the low affinity one ($K_i = 148$ nM), while MT₁ binding affinity was very low ($K_i = 1.4 \ \mu M$) resulting in the highest MT₂ selectivity of the whole series. These findings help to define the steric and electronic requirements of the hydrophobic binding pocket of MT₂ receptors accommodating the out-of-plane substituent of MT₂-selective melatonin receptor antagonists.

Experimental Section

General Methods. Melting points were determined using a capillary melting point apparatus (Gallenkamp, Sanyo) and are uncorrected. Column chromatography was carried out on silica gel 60 (0.063–0.200 mm) obtained from Merck. A Bruker AV-400 spectrometer was used to obtain ¹H NMR (400 MHz) and ¹³C NMR

(100 MHz) spectra. Proton chemical shifts are referenced to $CHCl_3$ (7.24 ppm) and DMSO- d_6 (2.55 ppm). Coupling constants (*J*) are given in hertz (Hz). Carbon chemical shifts are referenced to $CDCl_3$ (77.00 ppm) and DMSO- d_6 (39.50 ppm). The NMR resonances were assigned by means of HH-COSY, HMQC, and HMBC experiments. EI mass spectra were determined on Finnigan MAT 8200 and on ESI-microTOF spectrometers. IR spectra, recorded as ATR, were obtained by using a Biorad PharmalyzIR FT-IR instrument. Elemental analyses were performed by the Microanalytical Section of the Institute of Inorganic Chemistry, University of Würzburg. All reactions were carried out under an argon atmosphere. All chemicals were purchased from commercial suppliers and used directly without any further purification. The radioligand 2-[¹²⁵I]iodomelatonin was purchased from Perkin-Elmer (Shelton, CT).

General Procedure for the Synthesis of 2-(2,3-Dihydro-1*H*indol-1ylcarbonyl)-5-methoxy-1*H*-indoles 2a,b. A solution of the appropriate amine (1 equiv) in dry CH_2Cl_2 (5 mL) was added to a stirred solution of 5-methoxyindole-2-carboxylic acid (1 equiv) and EDCI·HCl (1.5 equiv) in dry CH_2Cl_2 (15 mL). The reaction mixture was stirred for 18 h at room temperature, extracted with 5 N hydrochloric acid (3 × 5 mL), washed with water (2 × 10 mL), and dried (Na₂SO₄). The organic layer was evaporated in vacuo, and the residue was recrystallized from isopropanol.

5-Methoxy-2-[(2-methyl-2,3-dihydro-1H-indol-1yl)carbonyl]-1H-indole (2a). Compound **2a** (1.02 g, 86%) was obtained from 5-methoxyindole-2-carboxylic acid (0.74 g) and 2-methylindoline as a pale-yellow powder, mp 174–176 °C. MS (EI): m/z (%) = 306 (M⁺, 48), 174 (55), 133 (100), 118 (49). Anal. (C₁₉H₁₈N₂O₂) C, H, N.

2-[(2,3-Dihydro-1*H***-indol-1yl)carbonyl]-5-methoxy-1***H***-indole (2b). Compound 2b** (0.45 g, 87%) was obtained from 5-methoxyindole-2-carboxylic acid (0.34 g) and indoline as a pale-yellow powder, mp 218–220 °C. MS (EI): m/z (%) = 292 (M⁺, 41), 174 (50), 119 (100). ¹H NMR (DMSO- d_6): δ 3.28 (t, 2H, J = 8.2), 3.82 (s, 3H), 4.54 (t, 2H, J = 8.2), 6.95 (dd, 1H, J = 8.8, 2.3), 7.08–7.12 (m, 2H), 7.12 (d, 1H, J = 2.3), 7.27 (dd, 1H, J = 7.6, 7.6), 7.35 (d, 1H, J = 7.3), 7.44 (d, 1H, J = 8.8), 8.22 (d, 1H, J = 7.8), 11.65 (br, 1H). ¹³C NMR (DMSO- d_6): δ 28.2, 49.6, 55.2, 102.1, 105.2, 113.1 (C-7), 115.3, 116.9, 123.8, 124.8, 126.9, 127.5, 131.0, 131.5, 132.3, 143.3, 153.8, 160.2. IR (cm⁻¹) ν = 3277, 3249, 1614, 1577. Anal. (C₁₈H₁₆N₂O₂) C, H, N.

General Procedure for the Synthesis of 1-[(2,3-Dihydro-1*H*-indol-1ylcarbonyl)-5-methoxy-1*H*-indole-3yl]-*N*,*N*-dimethylmethanamines 3a,b. Dimethylmethyleniminium iodide (1.3 equiv) was added to a solution of 2a,b (1.0 equiv) in dry CHCl₃ (50 mL). The reaction mixture was refluxed for 18 h, allowed to cool, and basified with 25% ammonia. The organic layer was separated, washed with water (4 × 15 mL), dried (Na₂SO₄), and evaporated under reduced pressure to afford the Mannich bases 3a,b. The crude products were pure enough to be used in the next step without further purification as indicated by ¹H NMR.

1-{5-Methoxy-2-[(2-methyl-2,3-dihydro-1*H*-indol-1ylcarbonyl)-1*H*-indol-3yl]}-*N*,*N*-dimethylmethanamine (3a). Compound 3a (0.64 g, 82%) was obtained from 2a (0.66 g).

1-[2-(2,3-Dihydro-1*H***-indol-1ylcarbonyl)-5-methoxy-1***H***-indol-3yl]-***N***,***N***-dimethylmethanamine (3b). Compound 3b (0.38 g, 79%) was obtained from 2b (0.40 g). ¹H NMR (CDCl₃): \delta 2.18 (s, 6H), 3.01 (t, 2H,** *J* **= 8.2), 3.59 (s, 2H), 3.83 (s, 3H), 4.08 (t, 2H,** *J* **= 8.2), 6.86 (dd, 1H,** *J* **= 8.8, 2.5), 6.95–7.03 (m, 2H), 7.16–7.19 (m, 3H), 7.19 (d, 1H,** *J* **= 2.5), 9.23 (br, 1H). ¹³C NMR (CDCl₃): \delta 27.7, 45.2, 49.6, 53.8, 55.7, 101.9, 112.5, 113.9, 114.8, 115.8, 123.9, 125.0, 127.2, 128.2, 131.2, 130.9, 132.8, 141.9, 154.3, 162.7.**

General Procedure for the Synthesis of [(2,3-Dihydro-1*H*-indol-1ylcarbonyl)-5-methoxy-1*H*-indole-3yl]acetonitriles 4a,b. Methyl iodide (1.2 equiv) was added to a solution of 3a,b (1.00 equiv) in dry CH₂Cl₂ (20 mL). The reaction mixture was stirred at room temperature for 1 h. The volatiles were removed under reduced pressure, and the residual ammonium salt was dissolved in dry acetonitrile (30 mL). Dicyclohexyl-18-crown-6 (0.10 g) and potassium cyanide (5 equiv) were added, and resulting reaction mixture



Figure 4. Competition of melatonin or 6e for 2- Γ^{125} I]iodomelatonin binding to MT₁ or MT₂ receptors expressed in CHO cells. Each data point represents the mean \pm standard error of three independent experiments performed in duplicate. Curves were fit by nonlinear regression analysis by one- or two-site fit to obtain affinity (K_i) values using the GraphPad Prism software.



Figure 5. Functional analysis of melatonin, **6b** (top graphs), and **6e** (bottom graphs) on forskolin-stimulated cAMP formation in CHO cells expressing either the MT_1 receptor or the MT_2 receptor. Each data point represents the mean \pm standard error of two to three independent experiments performed in duplicate. Curves were fit by nonlinear regression analysis by one- or two-site fit to obtain potency (IC₅₀) values using the GraphPad Prism software.

was heated at 80 °C for 1 h. The solvent was evaporated under vacuum and the residue was subjected to silica gel chromatography to give 4a,b.

{5-Methoxy-2-[(2-methyl-2,3-dihydro-1*H*-indol-1ylcarbonyl)-1*H*-indol-3-yl]}acetonitrile (4a). Compound 4a (0.40 g, 84%) was obtained from 3a (0.50 g) as a light-brown solid, mp 95–98 °C. Eluent: chloroform/ethyl acetate, 1:1. MS (ESI): m/z (%) = 346 (M + H⁺, 100). Anal. (C₂₁H₁₉N₃O₂) C, H, N.

[2-(2,3-Dihydro-1*H*-indol-1ylcarbonyl)-5-methoxy-1*H*-indol-3yl]acetonitrile (4b). Compound 4b (0.15 g, 61%) was obtained from 3b (0.26 g) as a pale-yellow solid, mp 199–201 °C. Eluent: chloroform/ethyl acetate, 1:9. MS (ESI): m/z (%) = 332 (MH⁺, 100). ¹H NMR (CDCl₃): δ 3.07 (t, 2H, J = 8.1), 3.85 (s, 5H), 4.14 (t, 2H, J = 8.1), 6.95 (dd, 1H, J = 8.8, 2.3), 7.00–7.03 (m, 2H), 7.11 (d, 1H, J = 2.3), 7.19–7.24 (m, 3H), 9.00 (br, 1H). ¹³C NMR (CDCl₃): δ 13.8, 27.9, 50.1, 55.8, 99.9, 106.4, 113.2, 115.7, 116.5, 117.3, 124.6, 125.4, 127.5, 126.7, 129.2, 130.8, 132.9, 141.6, 155.1, 161.1. IR (cm⁻¹) ν = 3280, 2832, 2197, 1614. Anal. (C₂₀H₁₇N₃O₂) C, H, N.

General Procedure for the Synthesis of Amides 6a–i. A solution of **4a–i** (1.0 equiv) in dry THF (5 mL) was added dropwise to a stirred suspension of LiAlH₄ (10.0 equiv) in dry diethyl ether (30 mL) at 0-5 °C. The reaction mixture was heated at 40 °C for 4 h. The reaction was quenched by a slow addition of saturated sodium sulfate solution at 0-5 °C. The formed precipitate was filtered off

and washed with THF (10 mL). The combined filtrate and washings were dried (Na₂SO₄), filtered, and evaporated under reduced pressure to give amines **5a,b,f**-**i** as viscous oils. The crude amines (1 equiv) were dissolved in dry CH₂Cl₂ (15 mL) and treated with triethylamine (3.5 equiv) and the appropriate acid anhydride (5.0 equiv for **6a**-**d** and **f**-**i**) or acid chloride (1 equiv for **6e**) at 0– 5 °C. The reaction mixture was stirred at ambient temperature for 18 h under an inert atmosphere. The solvent was evaporated under reduced pressure and the residue was purified by silica gel chromatography to give **6a**-**i**.

N-(2-{5-Methoxy-2-[(2-methyl-2,3-dihydro-1*H*-indol-1yl)methyl)]-1*H*-indole-3-yl}ethyl)acetamide (6a). Compound 6a (0.24 g, 75%) was obtained from 4a (0.29 g) as a yellow solid, mp 63–65 °C. Eluent: chloroform/methanol/ammonia, 10:1:0.1. MS (EI): m/z (%) = 377 (M⁺, 59), 291 (17), 245 (100), 186 (96), 174 (43), 118 (50). Anal. (C₂₃H₂₇N₃O₂) C, H, N.

N-(2-{5-Methoxy-2-[(2,3-dihydro-1*H*-indol-1yl)methyl)]-1*H*-indol-3-yl}ethyl)acetamide (6b). Compound 6b (0.095 g, 58%) was obtained from 4b (0.15 g) as a pale-yellow solid, mp 62–64 °C. Eluent: chloroform/ethyl acetate, 1:1. ¹H NMR (CDCl₃): δ 1.77 (s, 3H), 2.93–2.97 (m, 4H), 3.25–3.32 (m, 2H), 3.48 (m, 2H), 3.84 (s, 3H), 4.29 (s, 2H), 5.86 (br, 1H), 6.52 (d, 1H, *J* = 7.8), 6.75 (dd, 1H, *J* = 7.6, 7.3), 6.82 (dd, 1H, *J* = 8.8, 2.5), 7.00 (d, 1H, *J* = 2.3), 7.04 (dd, 1H, *J* = 7.8, 7.6), 7.12 (d, 1H, *J* = 7.3), 7.18 (d, 1H, *J* = 8.8), 8.45 (br, 1H). ¹³C NMR (CDCl₃): δ 23.1, 24.0, 28.5,

40.2, 46.2, 54.3, 55.9, 100.4, 108.3, 109.7, 111.7, 112.0, 114.9, 124.8, 127.5, 128.8, 130.5, 130.6, 132.8, 151.8, 154.1, 170.1. IR (cm⁻¹) ν = 3254, 2924, 1633 cm⁻¹. HRMS (ESI, pos) C₂₂H₂₅N₃-O₂·Na⁺: *m/z* calcd 386.1844, *m/z* found 386.1839. Anal. (C₂₂H₂₅-N₃O₂) C,H,N.

N-(2-{5-Methoxy-2-[(2,3-dihydro-1*H*-indol-1yl)methyl)]-1*H*-indol-3-yl}ethyl)butanamide (6c). Compound 6c (0.15 g, 64%) was obtained from 4a (0.20 g) as a light-brown solid, mp 54–56 °C. Eluent: chloroform/methanol/ammonia, 10:1:0.1. HRMS (ESI, pos) $C_{25}H_{31}N_3O_2 \cdot Na^+$: *m/z* calcd 328.2314, *m/z* found 328.2309. Anal. ($C_{25}H_{31}N_3O_2$) C, H, N.

N-(2-{5-Methoxy-2-[2,3-dihydro-1*H*-indol-1yl)methyl)]-1*H*-indol-3-yl}ethyl)propanamide (6d). Compound 6d (0.11 g, 88%) was obtained from 4b (0.11 g) as a pale-yellow solid, mp 55–57 °C. Eluent: chloroform/ethyl acetate, 1:1. MS (EI): m/z (%) = 377 (M⁺, 54), 259 (90), 203 (87), 186 (100), 174 (56). Anal. (C₂₃H₂₇N₃O₂) C, H, N.

N-(2-{5-Methoxy-2-[(2,3-dihydro-1H-indol-1yl)methyl)]-1H-indol-3-yl}ethyl)cyclobutanecarboxamide (6e). Compound 6e (0.04 g, 22%) was obtained from 4b (0.15 g) as a beige solid, mp 66-68 °C, and the reaction time was only 1 h at 0 °C. Eluent: chloroform/ethyl acetate, 1:1. MS (EI): m/z (%) = 403 (M⁺, 34), 285 (61), 203 (65), 186 (66), 118 (100). ¹H NMR (CDCl₃): δ 1.70-1.91 (m, 2H), 1.95-2.03 (m, 2H), 2.13-2.23 (m, 2H), 2.73-2.82 (m, 1H), 2.93-2.97 (m, 4H), 3.28 (t, 2H, J = 8.1), 3.49 (m, 2H), 3.84 (s, 3H), 4.31 (s, 2H), 5.58 (br, 1H), 6.49 (d, 1H, J =7.8), 6.73 (dd, 1H, J = 7.6, 7.3), 6.81 (dd, 1H, J = 8.8, 2.5), 6.99 (d, 1H, J = 2.3), 7.04 (dd, 1H, J = 7.8, 7.6), 7.12 (d, 1H, J = 7.3), 7.19 (d, 1H, J = 8.8), 8.26 (br, 1H). ¹³C NMR (CDCl₃): δ 18.1, 24.2, 25.3, 28.6, 39.9, 40.0, 45.9, 54.3, 56.0, 100.5, 107.8, 109.5, 111.6, 111.8, 119.0, 124.8, 127.5, 128.9, 130.4, 130.5, 133.2, 152.2, 154.1, 174.9. IR (cm⁻¹) ν = 3265, 2931, 1635, 1484. Anal. (C₂₅H₂₉N₃O₂) C, H, N.

N-(2-{5-Methoxy-2-[(5-methoxy-2,3-dihydro-1*H*-indol-1-yl)methyl)]-1*H*-indole-3-yl}ethyl)acetamide (6f). Compound 6f (0.08 g, 37%) was obtained from 4f (0.20 g) as a yellow solid, mp 158–160 °C, Eluent: chloroform/methanol/ammonia, 10:1:0.1. MS (EI): m/z (%) = 393 (M⁺, 17), 245 (24), 186 (25), 149 (79), 134 (100). Anal. (C₂₃H₂₇N₃O₃) C, H, N.

N-(2-{5-Methoxy-2-[(5-methyl-2,3-dihydro-1*H*-indol-1yl)methyl)]-1*H*-indol-3-yl}ethyl)acetamide (6g). Compound 6g (0.15 g, 49%) was obtained from 4g (0.28 g) as a pale-yellow solid, mp 65– 68 °C. Eluent: ethyl acetate. MS (EI): m/z (%) = 377 (M⁺, 41), 245 (68), 203 (43), 186 (63), 132 (100). Anal. (C₂₃H₂₇N₃O₂) C, H, N.

N-(2-{2-[(5-Bromo-2,3-dihydro-1*H*-indol-1yl)methyl)]-5-methoxy-1*H*-indole-3-yl}ethyl)acetamide (6h). Compound 6h (0.13 g, 80%) was obtained from 4h (0.15 g) as a pale-yellow solid, mp 149–150 °C. Eluent: ethyl acetate. MS (EI): m/z (%) = 442, 444 (M + H⁺, 100). Anal. (C₂₂H₂₄BrN₃O₂) H. C: calcd, 59.74; found, 59.10. N: calcd, 9.50; found, 9.05.

N-{2-[5-Methoxy-2-(pyrrolidin-1ylmethyl)-1*H*-indol-3yl]ethyl}acetamide (6i). The reaction time for the acetylation of 5i was reduced to 30 min. Subsequently, 2 M NaOH (5 mL) was added and stirring was continued for 3 h. After addition of water (100 mL) the reaction mixture was extracted with CH₂Cl₂ (2 × 30 mL). The combined organic layers were washed with water and dried over MgSO₄. The solvent was evaporated under reduced pressure, and the residue was purified by silica gel chromatography (ethyl acetate). Compound 6i (0.13 g, 39%) was obtained from 4i (0.30 g) as a colorless solid, mp 155–156 °C. MS (EI): m/z (%) = 316 (M + H⁺, 100). Anal. (C₁₈H₂₅N₃O₂) C, H, N.

Molecular Modeling. The 3D structures of both enantiomers of **1b** were obtained by conformational analysis using the Cartesian method of the steric energy minimization program GLOBAL-MMX implemented in Pcmodel, version 9.0,⁴⁵ and further optimized by means of semiempirical PM3 calculations of HyperChem, version 7.1.⁴⁶ For the superposition of **1b** onto UCM454 (HyperChem), only the benzene rings bearing the methoxy substituent were considered.

Competition Binding Analysis. All synthesized compounds were tested for their binding affinity and selectivity for each of the melatonin receptor subtypes MT₁ and MT₂ using competition binding analysis.

Briefly, cells expressing the human MT_1 or MT_2 melatonin receptor (MT₁-CHO, MT₂-CHO) were grown to confluence on 10 cm cell culture plates until they reached approximately 80% confluence. Next, cells were washed, lifted, and added to tubes containing 80-100 pM 2-[¹²⁵]]iodomelatonin in the absence (total binding) or presence of melatonin (1 fM to 1 μ M) or the test compounds (1 pM to 1 μ M). The mixtures were incubated for 1 h at 25 °C, and then the reaction was terminated following the addition of cold Tris-HCl solution (50 mM, pH 7.4). The mixtures were filtered through glass fiber filters (Schleicher and Schuell, Keene, NH) saturated in polyethylenimine 0.5% solution (v/v). Radioactive counts were counted using a γ counter. Data points were fit by one- or two-site nonlinear regression analysis based on the lowest residual sum of squares (GraphPad Prism), and affinity constants (K_i) were calculated. The affinity values (K_i) of each test compound were compared between receptors to obtain selectivity profiles of each. Compounds whose affinity values were in the subnanomolar range and whose selectivity profile was greater than 50 were subjected to functional analysis as described below.

Cyclic AMP Assays. The cAMP accumulation assays were carried out using Enzyme Immuno Antibody (EIA) kit according to the manufacturer's directions. Stable CHO cell lines expressing human MT₁ or human MT₂ receptors were cultured on 10 cm plates in F-12 media containing 10% FBS and 1% pen/strep until they were 70-80% confluent, after which the cells were lifted and plated in 24-well plates. The following day, the cells were incubated in serum-free media containing one of the following treatment groups for 20 min at 37 °C: (a) 30 μ M rolipram alone (basal); (b) 30 μ M rolipram and 100 μ M forskolin (maximal accumulation); (c) 30 μ M rolipram, 100 μ M forskolin, and melatonin (in concentrations ranging from 10^{-14} to 10^{-7} M); (d) 30 μ M rolipram, 100 μ M forskolin, and **6b** (in concentrations ranging from 10^{-13} to 10^{-8} M) or **6e** (in concentrations ranging from 10^{-14} to 10^{-4} M). Co-incubation of melatonin (from 10^{-14} to 10^{-7} M) along with 10 or 100 nM 6b was also performed to determine the competitive nature of 6b at MT1 or MT2 receptors. Cyclic AMP accumulation was expressed as a percentage of forskolin response within each group. Curves were fit using nonlinear regression analysis (one-site or two-site), and potency (IC₅₀) values were calculated using the commercially available software (GraphPad Prism; GraphPad Prism, Inc., San Diego, CA).

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Supporting Information Available: Experimental data for compounds **7–9**, **4f–j**, and **6j,k**; ¹H NMR, ¹³C NMR, and IR data for compounds **7–9**, **2a**, **3a**, **4a,f–j**, **6a,c,d,f–k**; elemental analysis for all novel compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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